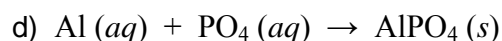
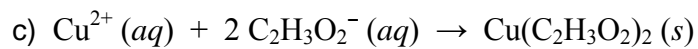
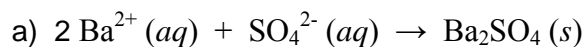
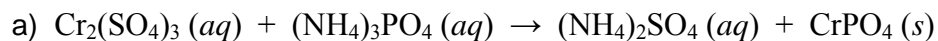


Writing Net Ionic Equations

1. Identify the mistakes in each of the following incorrect net ionic equations, and then correct them.



2. Write net ionic equations for each of these reactions. In some cases you will first have to determine the products of the reaction and/or balance the chemical equation. One of the reactions below does not occur (i.e., no products form).



b) sodium bromide solution + lead (II) nitrate solution

c) solid sodium carbonate is added to a solution of silver nitrate

d) strontium sulfide solution + calcium bromide solution

e) ammonium hydroxide solution + tin (IV) sulfate solution