## **Some Common Polyatomic Ions to Know**

Cations
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Ammonium ion	$NH_4^+$	
Hydronium ion	H <sub>3</sub> O <sup>+</sup>	(In solution; no common compounds)
Mercury(I)	$Hg_2^{2+}$	

## **Anions**

Acetate ion	$C_2H_3O_2^-$	(Also written CH <sub>3</sub> CO <sub>2</sub> or CH <sub>3</sub> COO)
Carbonate ion	CO <sub>3</sub> <sup>2-</sup>	

Hydrogen carbonate ion  $HCO_3^-$  (Formerly called bicarbonate)

Hypochlorite ion CIO<sup>-</sup> CIO<sub>2</sub> Chlorite ion CIO<sub>3</sub> Chlorate ion Perchlorate ion CIO<sub>4</sub> CrO<sub>4</sub> 2-Chromate ion Cr<sub>2</sub>O<sub>7</sub> 2-Dichromate ion Cyanate ion  $OCN^-$ Thiocyanate ion SCN<sup>-</sup>

Thiocyanate ion  $SCN^-$ Cyanide ion  $CN^-$ Hydroxide ion  $OH^-$ Nitrite ion  $NO_2^-$ Nitrate ion  $NO_3^-$ Oxalate ion  $C_2O_4^{\ 2^-}$ Permanganate ion  $MnO_4^-$ Peroxide  $O_2^{\ 2^-}$ Phosphate ion  $PO_4^{\ 3^-}$ 

Hydrogen phosphate ion  $HPO_4^{-2-}$  (Also called monohydrogen phosphate)

Dihydrogen phosphate ion  $H_2PO_4^-$ Sulfite ion  $SO_3^{2-}$ 

Hydrogen sulfite ion  $HSO_3^-$  (Formerly called bisulfite)

Sulfate ion  $SO_4^{2^-}$ 

Hydrogen sulfate ion  $HSO_4^-$  (Formerly called bisulfate)

Thiosulfate ion  $S_2O_3^{2-}$