

## Elements and Operations

- ☞ A **symmetry element** is an imaginary geometrical construct about which a symmetry operation is performed.
- ☞ A **symmetry operation** is a movement of an object about a symmetry element such that the object's orientation and position before and after the operation are indistinguishable.
- ✓ A symmetry operation carries every point in the object into an **equivalent** point or the **identical** point.

## Point Group Symmetry

- ☞ All symmetry elements of a molecule pass through a central point within the molecule.
- ☞ The more symmetry operations a molecule has, the higher its symmetry is.
- ☞ Regardless of how many or few symmetry operations a molecule possesses, all are examples of one of five types.

Operation	Element	Element Construct
Identity, $E$	The object	N/A
Proper rotation, $C_n$	Proper axis, Rotation axis	line
Reflection, $\sigma$	Mirror plane, Reflection plane	plane
Inversion, $i$	Inversion center, Center of symmetry	point
Rotation-reflection Improper rotation, $S_n$	Improper axis, alternating axis	line

## Proper Rotation, $C_n$

- ☞ If a molecule has rotational symmetry, rotation by  $2\pi/n = 360^\circ/n$  brings the object into an equivalent position.
- ✓ The value of  $n$  is the **order** of an  $n$ -fold rotation.
- ✓ If the molecule has one or more rotational axes, the one with the highest value of  $n$  is the **principal axis of rotation**.

Fig. 1.1 Successive  $C_4$  clockwise rotations of a planar  $MX_4$  molecule about an axis perpendicular to the plane of the molecule ( $X_A = X_B = X_C = X_D$ ).

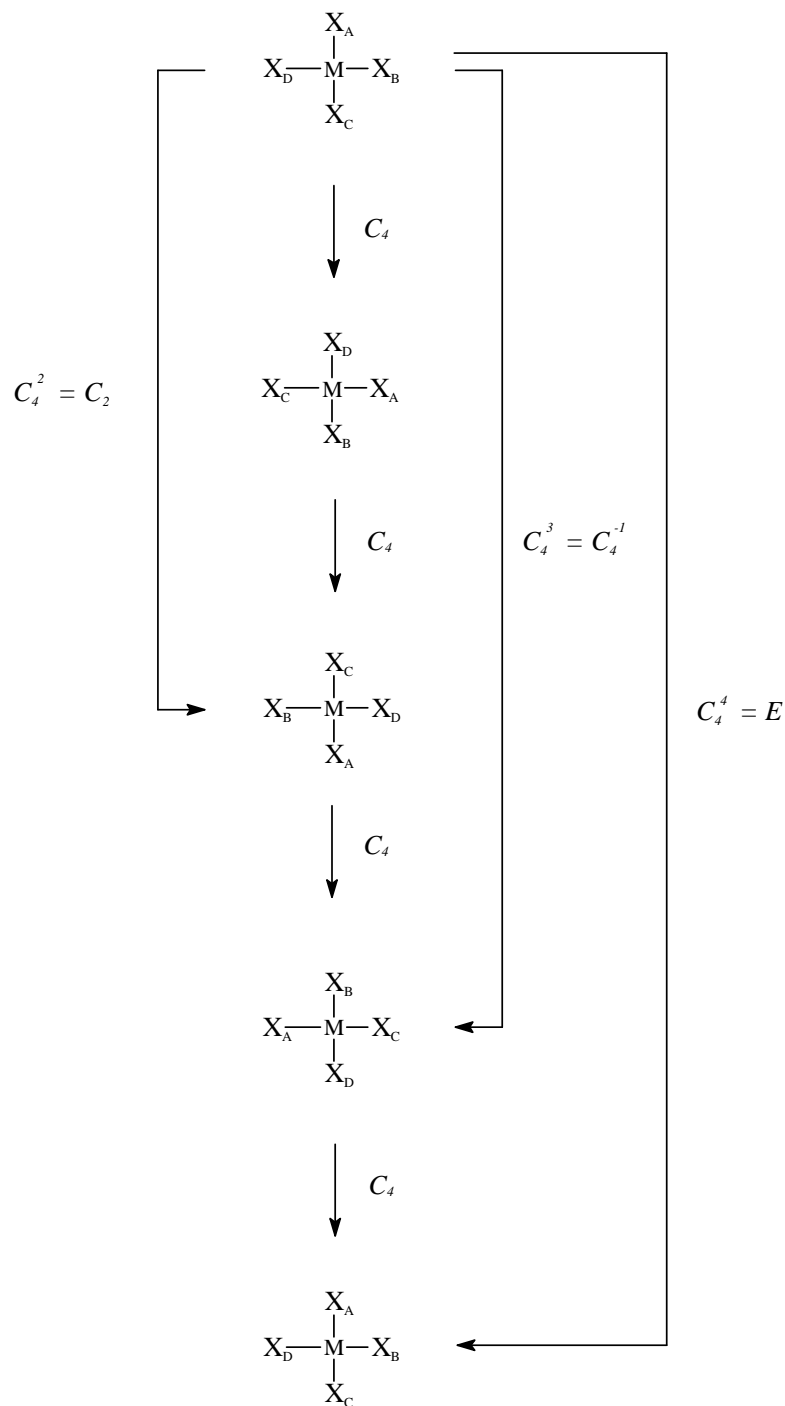
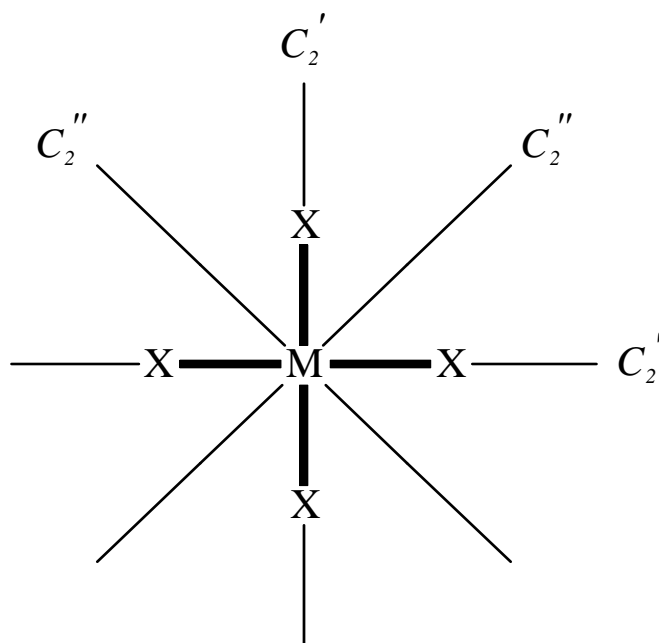


Fig. 1.2 The  $C_2'$  and  $C_2''$  axes of a planar  $MX_4$  molecule.



## General Relationships for $C_n$

$$C_n^n = E$$

$$C_n^{n/2} = C_2 \quad (n \text{ even})$$

$$C_n^{n-1} = C_n^{-1}$$

$$C_n^{n+m} = C_n^m \quad (m < n)$$

- ✓ Every  $n$ -fold rotational axis has  $n-1$  associated operations (excluding  $C_n^n = E$ ).

## Reflection, $\sigma$

- ☞ For every point a distance  $r$  along a normal to a mirror plane there exists a point at  $-r$ .

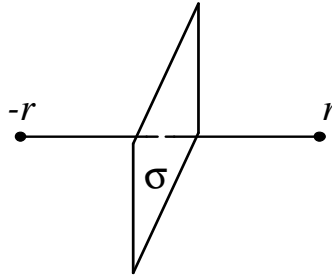
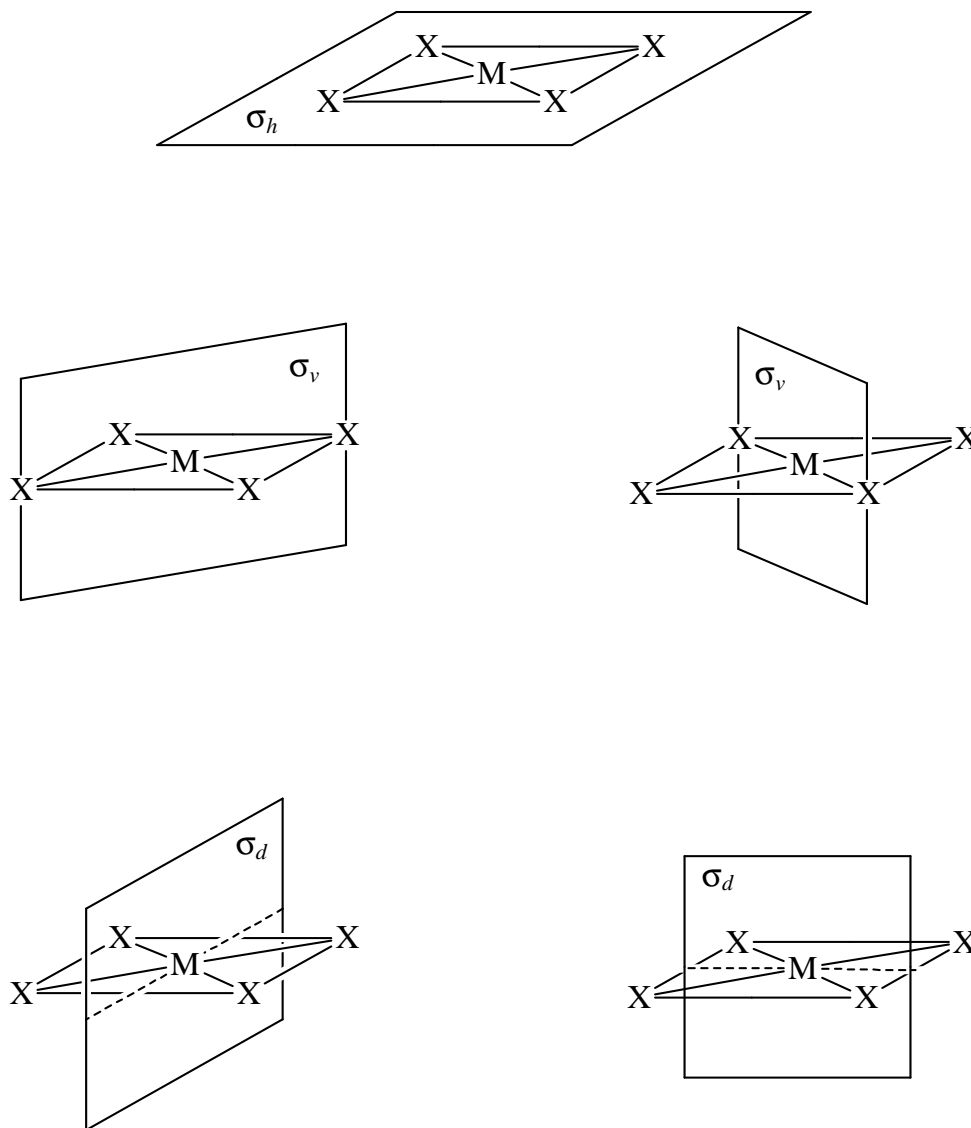


Fig. 1.3 Two points, equidistant from a mirror plane  $\sigma$ , related by reflection.

- ✓ For a point  $(x, y, z)$ , reflection across a mirror plane  $\sigma_{xy}$  takes the point into  $(x, y, -z)$ .
- ✓ Each mirror plane has only one operation associated with it, since  $\sigma^2 = E$ .

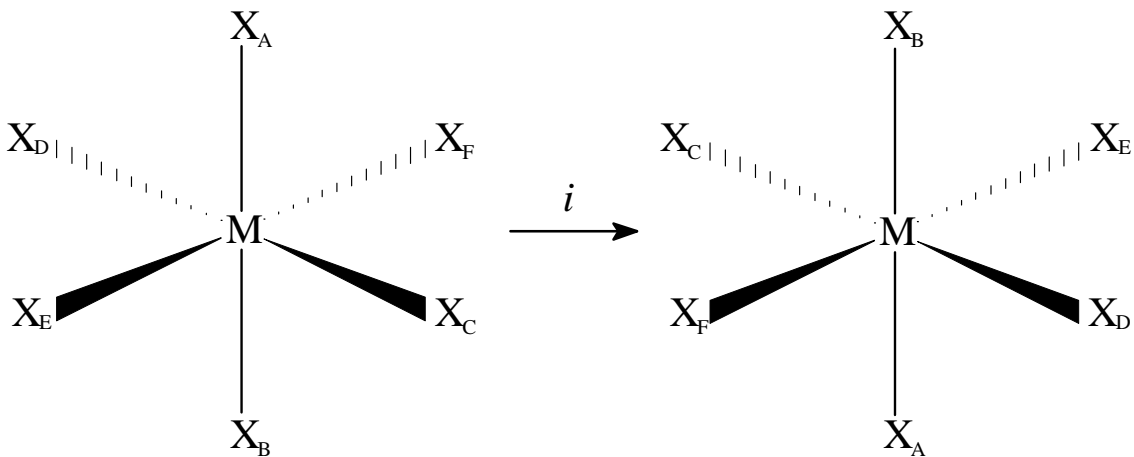
**Horizontal, Vertical, and Dihedral Mirror Planes**  
Fig. 1.4 Mirror planes of a square planar molecule  $\text{MX}_4$ .



## Inversion, $i$

- ☞ If inversion symmetry exists, for every point  $(x,y,z)$  there is an equivalent point  $(-x,-y,-z)$ .
- ✓ Each inversion center has only one operation associated with it, since  $i^2 = E$ .

Fig. 1.6 Effect of inversion ( $i$ ) on an octahedral  $MX_6$  molecule ( $X_A = X_B = X_C = X_D = X_E = X_F$ ).



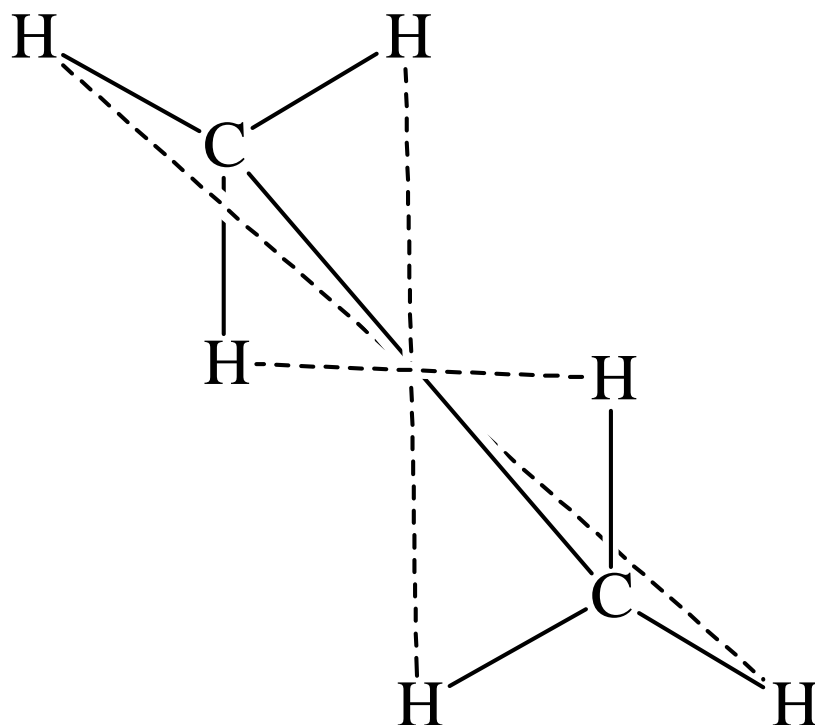


Fig. 1.7 Ethane in the staggered configuration. The inversion center is at the midpoint along the C-C bond. Hydrogen atoms related by inversion are connected by dotted lines, which intersect at the inversion center. The two carbon atoms are also related by inversion.

## Rotation-Reflection (Improper Rotation), $S_n$

☞  $S_n$  exists if the movements  $C_n$  followed by  $\sigma_h$  (or vice versa) bring the object to an equivalent position.

✓ If both  $C_n$  and  $\sigma_h$  exist, then  $S_n$  must exist.

Example:  $S_4$  collinear with  $C_4$  in planar  $MX_4$ .

✓ Neither  $C_n$  nor  $\sigma_h$  need exist for  $S_n$  to exist.

Example:  $S_4$  collinear with  $C_2$  in tetrahedral  $MX_4$ .

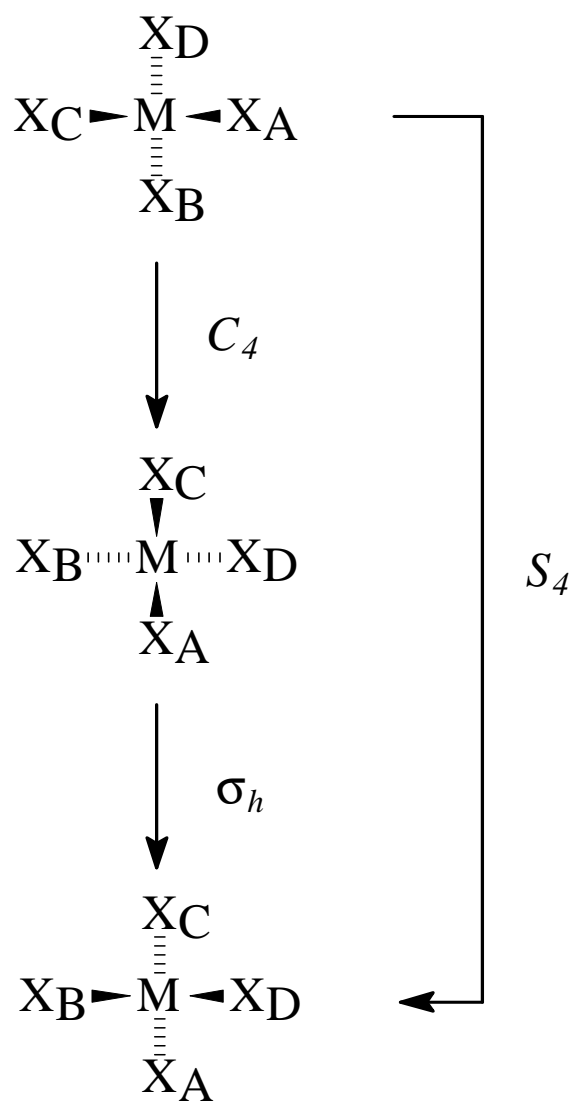


Fig. 1.8  $S_4$  improper rotation of a tetrahedral  $MX_4$  molecule ( $X_A = X_B = X_C = X_D$ ). The improper axis is perpendicular to the page. Rotation is arbitrarily taken in a clockwise direction. Note that neither  $C_4$  nor  $\sigma_h$  are genuine symmetry operations of tetrahedral  $MX_4$ .

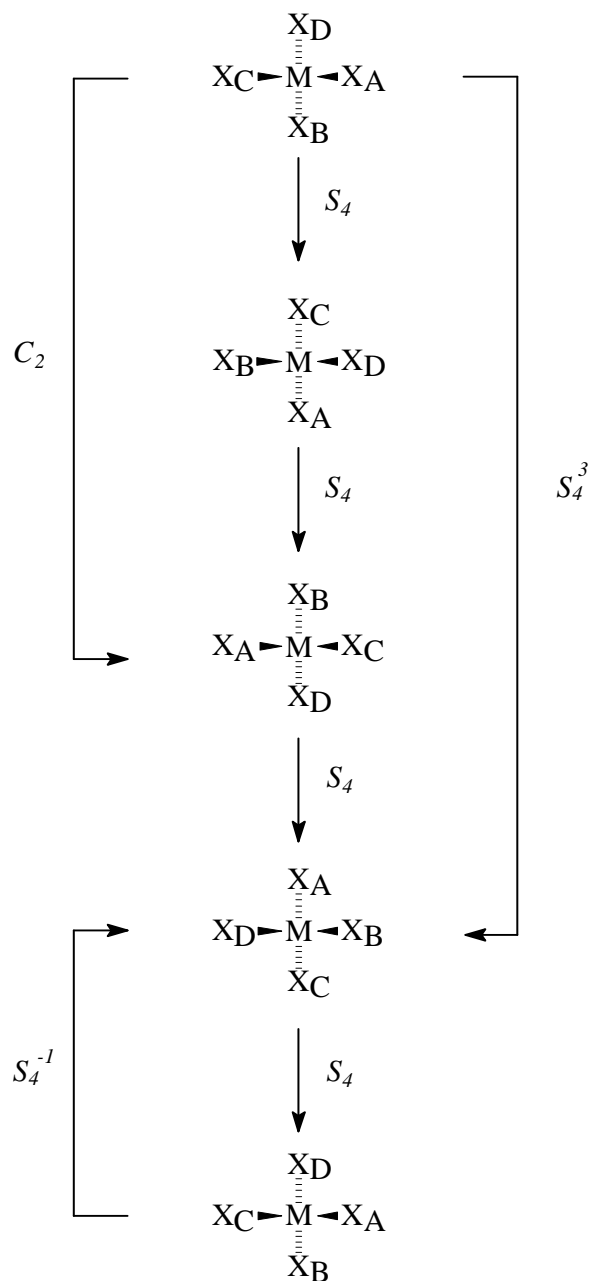


Fig. 1.9 Successive  $S_4$  operations on a tetrahedral  $MX_4$  molecule ( $X_A = X_B = X_C = X_D$ ). Rotations are clockwise, except  $S_4^{-1}$ , which is equivalent to the clockwise operation  $S_4^3$ .

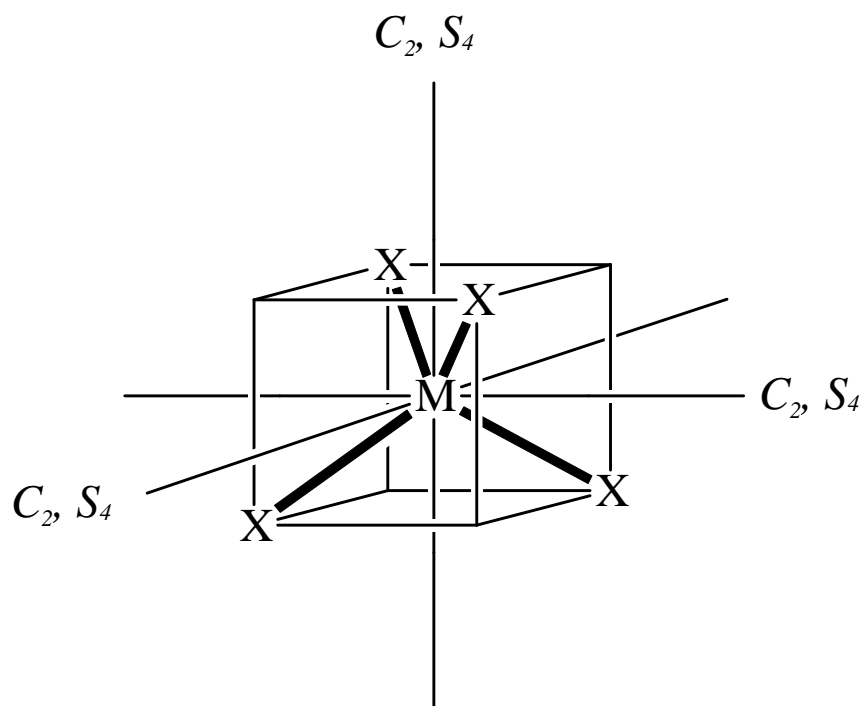
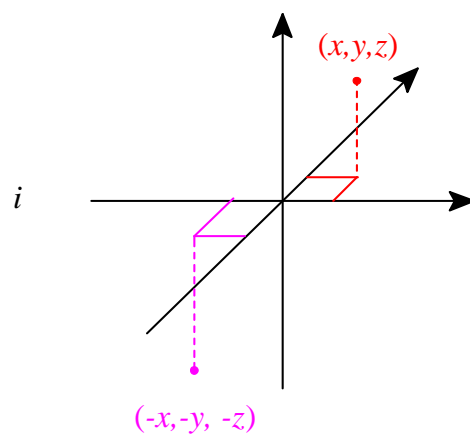
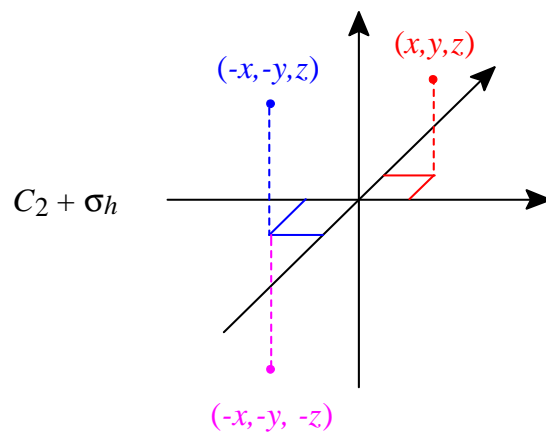


Fig. 1.10 A tetrahedral  $\text{MX}_4$  molecule inscribed in a cube. A  $C_2$  axis, collinear with an  $S_4$  axis, passes through the centers of each pair of opposite cube faces and through the center of the molecule.

## Non-Genuine $S_n$ Operations:

☞  $S_1 = \sigma$

☞  $S_2 = i$



## General Relations of Improper Axes

Equivalences of successive  $S_n$  operations:

- ✓ If  $n$  is even,  $S_n^n = E$
- ✓ If  $n$  is odd,  $S_n^n = \sigma$  and  $S_n^{2n} = E$
- ✓ If  $m$  is even,  $S_n^m = C_n^m$  when  $m < n$  and  $S_n^m = C_n^{m-n}$  when  $m > n$
- ✓ If  $S_n$  with even  $n$  exists, then  $C_{n/2}$  exists
- ✓ If  $S_n$  with odd  $n$  exists, then both  $C_n$  and  $\sigma$  perpendicular to  $C_n$  exist.

## Examples

☞ Find all symmetry elements and operations in the following:

