

# Osorb: A Novel Water Remediation Technique

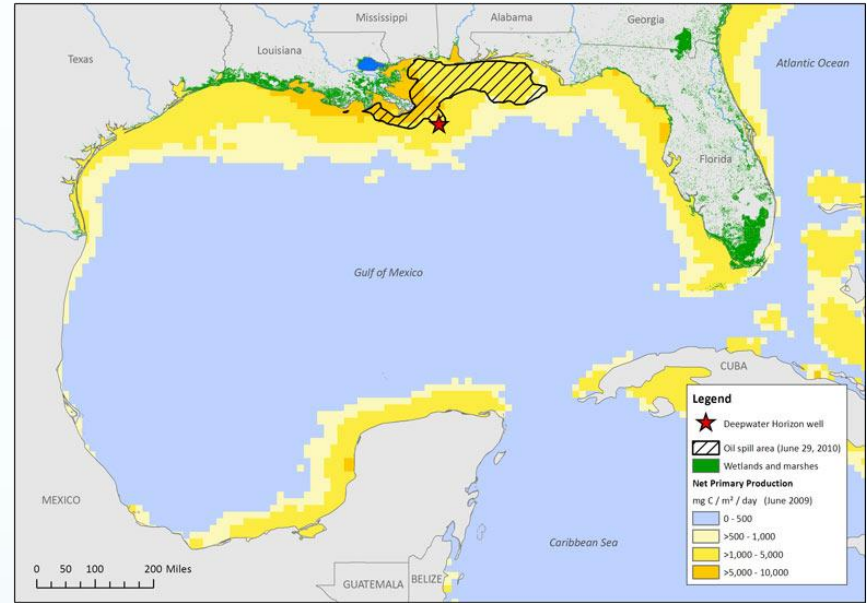
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6 December 2011

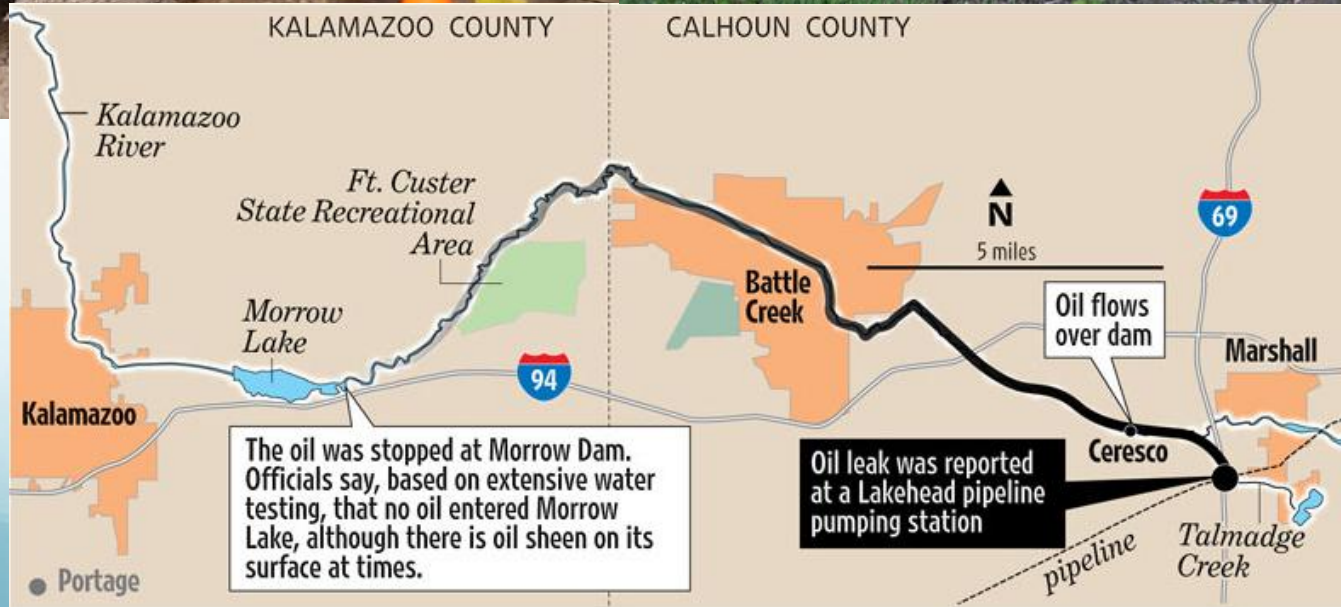
# Outline

- Applications
- Background of Oil Spill Cleanup
- Absorbent Material
- Osorb Synthesis
- Swelling Capabilities
- Green Application
- Conclusion

# Deepwater Horizon Oil Spill



# Talmadge Creek Oil Spill



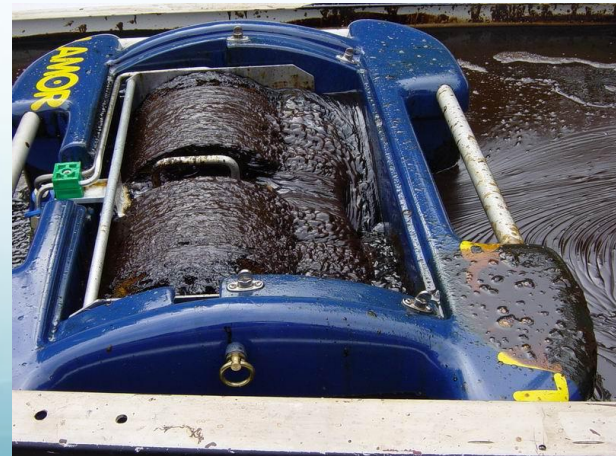


# Cleanup Methods

## Containment and Diversion



## Mechanical removal



# Cleanup Methods

Dispersants/ Chemical Treating Agents



In Situ Burning



# Characteristics of a good absorbent

- Hydrophobicity
- Oleophilicity
- High uptake capacity
- High rate of uptake
- Retention over time
- Oil recovery from absorbents
- Reusability and Biodegradability of absorbents

# Types of Absorbents

## Synthetic Organic Products

- Polypropylene
- Polyurethane

## Natural Products

- Rice straw
- Wool Fiber
- Peat moss wood
- Milkweed Floss



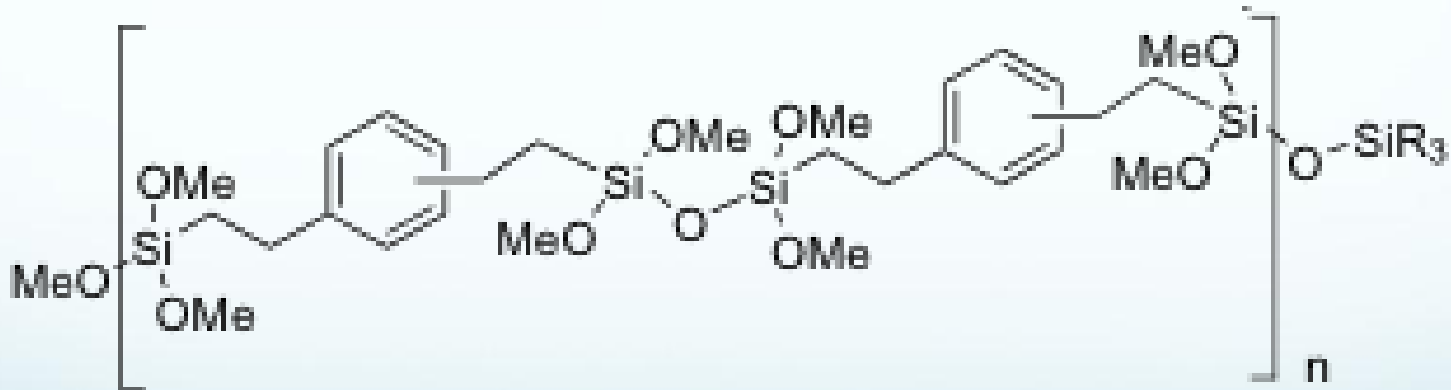
# Types of Absorbents

## Mineral Products

- Silica Aerogels
- Zeolites
- Organophilic Clays
- Osorb

# What is Osorb?

- Swellable Organically Modified Silica (SOMS)
- A sol-gel that after drying can swell in a few seconds when placed in an organic solvent



# Attributes

- Rate of swelling is mass transport limited
- Uptake of absorbates generates forces  $>100\text{N/g}$
- The swelling is completely reversible
- Absorption is non-selective
- Material is Hydrophobic
- Swelling and absorption is driven by the release of stored tensile force

# Synthesis of a Sol-Gel

## Step 1: Mixing /Gelation

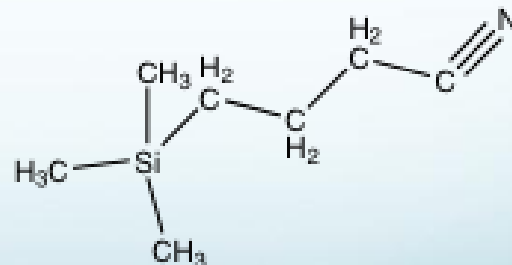


0.144 mol bis(trimethoxysilyl)benzene added to 220ml Acetone

1.5 mL of 1.0M tetrabutylammonium fluoride in 7.8 mL water

## Step 2: Aging and Rinsing

## Step 3: Derivatization



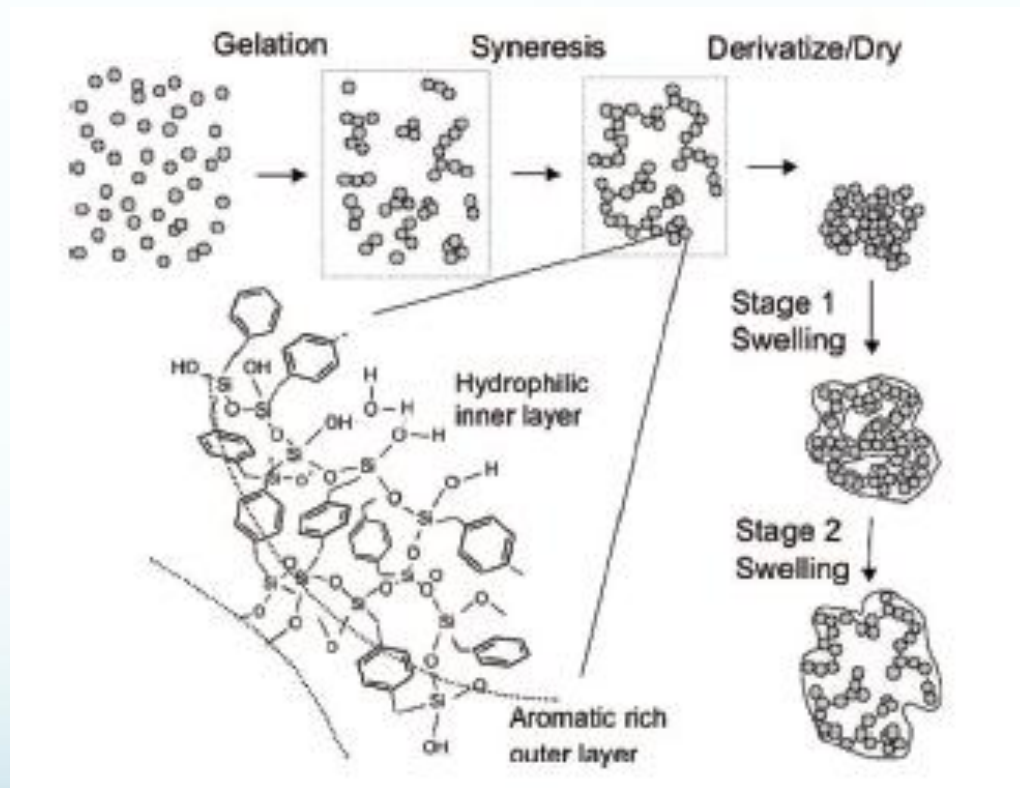
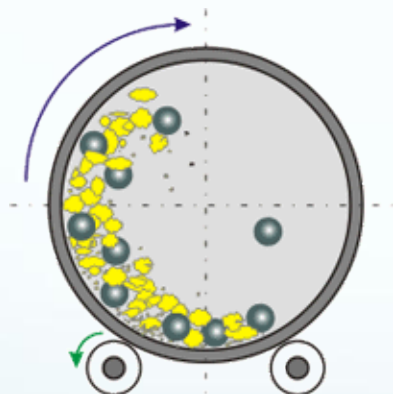
5% v/v cyanopropyltrimethylsilane in Acetonitrile



# Synthesis of a Sol-Gel

Step 4: Rinse and Dry

Step 5: Grind



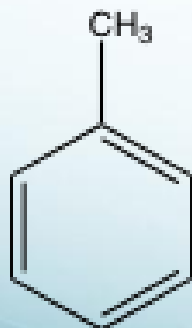
# Swelling Capability



# Absorption of Toluene

| Concentration (ppm) | Percent extraction** | Partition coefficient*** / 10 <sup>3</sup> |
|---------------------|----------------------|--|
| 25                  | 99.8                 | 285  |
| 55                  | 98.2                 | 21.8                                       |
| 100                 | 95.9                 | 9.4  |
| 210                 | 96.1                 | 9.8  |
| 320                 | 94.4                 | 6.7  |
| 420                 | 91.9                 | 4.5  |
| 530                 | 89.6                 | 3.4  |

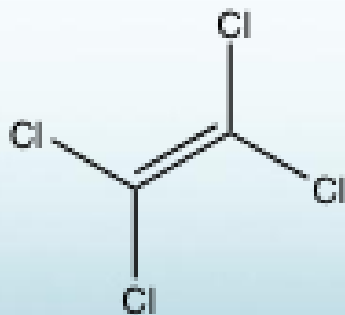
μg TCE abs/mg SOMS



11  
21  
40  
82  
120  
156  
190

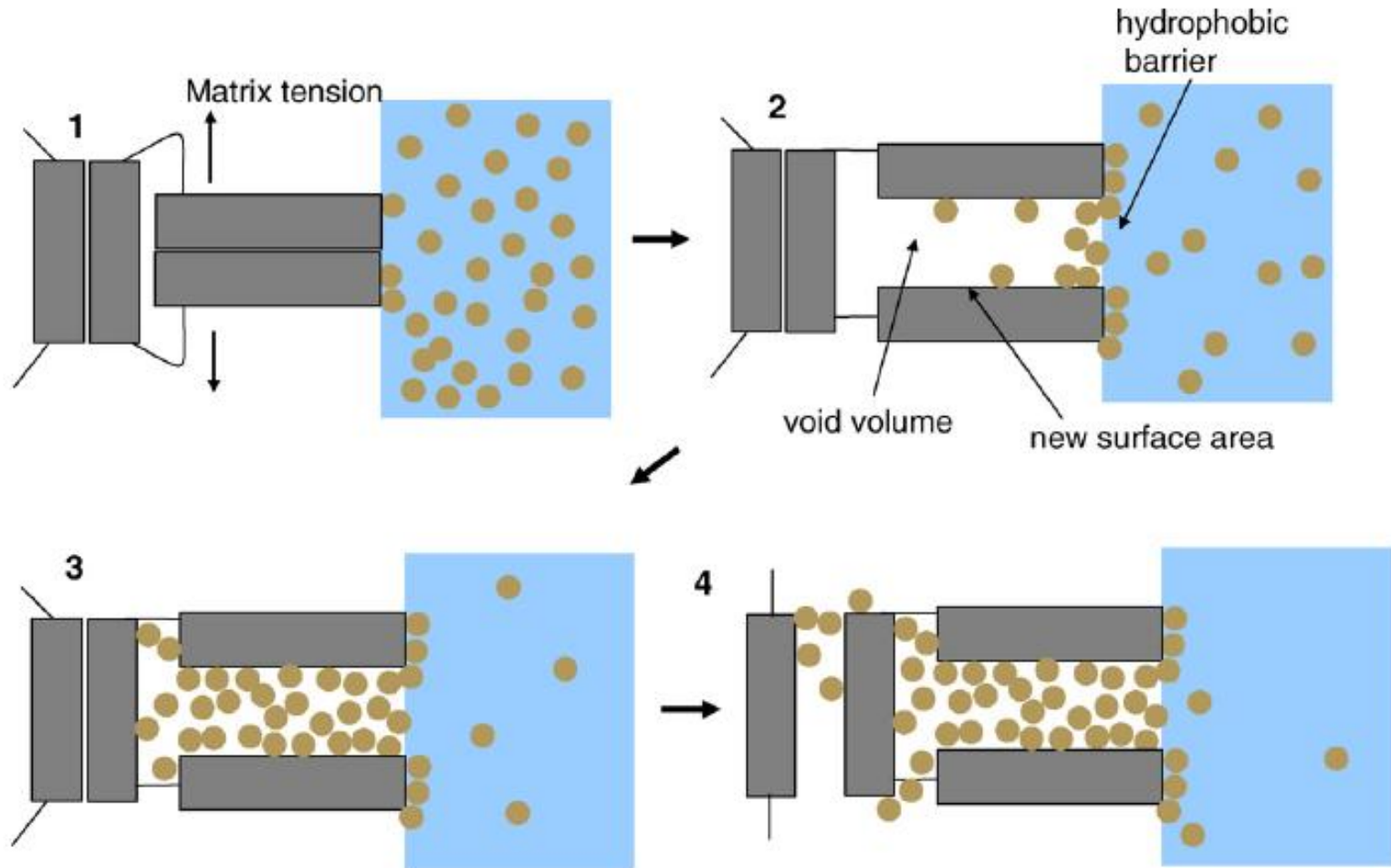
# Absorption of Perchloroethylene

| Concentration (ppm) | Partition coefficient/ $10^3$ | $\mu\text{g PCE abs/mg SOMS}$ |
|---------------------|-------------------------------|-------------------------------|
| 1.0                 | $4.0 \pm 0.8$                 | 0.2                           |
| 8.0                 | $6.3 \pm 0.8$                 | 1.7                           |
| 30                  | $21 \pm 1$                    | 6.2                           |
| 70                  | $19 \pm 4$                    | 13.6                          |
| 145                 | $16 \pm 2$                    | 28.3                          |





# Triggered Matrix Expansion



# Absorption of Polar Organics

Absorption data for polar organics in water<sup>\*</sup>.

| Concentration (ppm) | Percent extraction <sup>**</sup> |                   |                   |                   |
|---------------------|----------------------------------|-------------------|-------------------|-------------------|
|                     | MTBE                             | 1-Butanol         | 1,4-Dioxane       | Acetone           |
| 50                  | 48                               | 21                | 33                | 23                |
| 100                 | 52                               | 21                | 33                | 15                |
| 200                 | 53                               |                   | 30                | 25                |
| 500                 | 56                               | 43                | 18                | 5                 |
| 1000                | 46                               |                   | 24                | 5                 |
| 2000                | 32                               | 25                |                   |                   |
| 10,000              | 33                               | 26                | 33                | 17                |
| 20,000              | 38                               | 18                | 49 <sup>***</sup> | 13 <sup>***</sup> |
| 50,000              | 34 <sup>***</sup>                | 24 <sup>***</sup> |                   |                   |

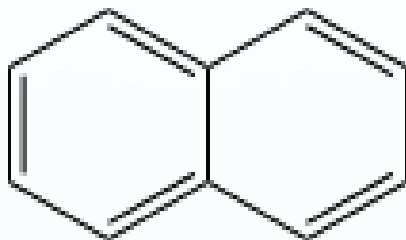
<sup>\*</sup> Mass SOMS/volume solution = 0.5% (w/v). Temperature = 25 °C.

<sup>\*\*</sup> Error <10% ( $n = 3$ ).

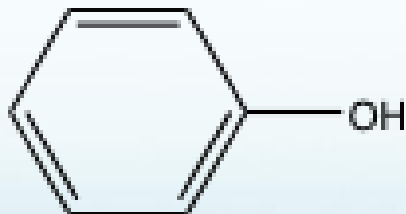
<sup>\*\*\*</sup> Visible swelling noted.

# Absorption of Other Organic Species

- Naphthalene – Solid at standard state



- Phenol – exists in a deprotonated anionic form at environmental pH



# Regeneration

- Tested by heating the SOMS that had absorbed either TCE or PCE
- Results: - Swelling capacity remained consistent (3.1mL)

TCE and PCE binding affinity after thermal regeneration<sup>\*</sup>.

| Number of times used | Dissolved contaminant <sup>**</sup> | Regeneration temperature (°C) | Partition coefficient <sup>***</sup> /10 <sup>3</sup> |
|----------------------|-------------------------------------|-------------------------------|---|
| 0                    | TCE                                 | n/a <sup>*</sup>              | 1.7 ± 0.7   |
| 1                    | TCE                                 | 60                            | 0.8 ± 0.2   |
| 2                    | TCE                                 | 60                            | 0.9 ± 0.4   |
| 1                    | TCE                                 | 110                           | 2.2 ± 0.5   |
| 2                    | TCE                                 | 110                           | 1.1 ± 0.2   |
| 0                    | PCE                                 | n/a                           | 11 ± 2  |
| 2                    | PCE                                 | 110                           | 9 ± 3   |

<sup>\*</sup> Mass SOMS/volume H<sub>2</sub>O = 0.5% (w/v). Temperature = 25 °C.

<sup>\*\*</sup> TCE concentration = 1200 ppm; PCE concentration = 15 ppm.

<sup>\*\*\*</sup> n = 3 for all measurements.



# Absorption from Natural Waters

- Salt does not inhibit absorption
- Water/Topsoil mixture does not absorb as well as pure water mixture
- Leaves behind water, proteins, genetic material, and inorganic salts

# How is this green?

1. Waste Prevention
2. Atom Economy
3. Less Hazardous Chemical Process
4. Designing Safer Chemicals
5. Safer Solvents and Auxiliaries
6. Energy Efficiency
7. Renewable Feedstocks
8. Reduce Derivatives
9. Catalysis
10. Design for Degradation
11. Real-time Analysis for Pollution Prevention
12. Safer Chemistry for Accident Prevention

# Conclusion

- SOMS are highly effective in absorbing organic species from water
- SOMS can be tailored to target specific contaminants
- Work is being done to explore more uses of Osorb

# References

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